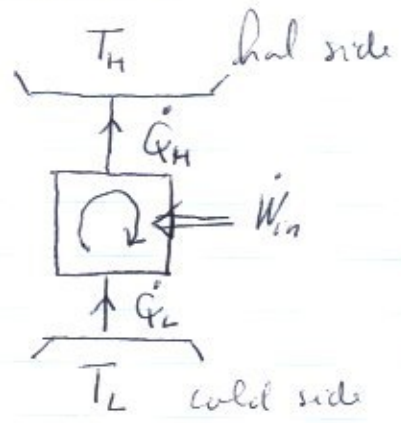


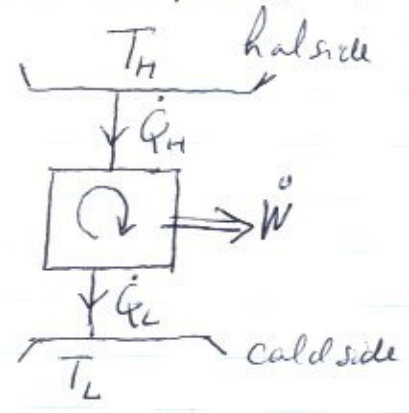
S 7.1, 7.2 Black box
Generic refrigerator,
freezer, AC, ...



Generic heat pump
producing heat

< like fridge >

Generic heat engine (car
engine, steam power plant, ...)



T_H : high temperature side temperature
 T_L : low " " "

Note: we are dropping our convention that heat in is positive for Q_H, Q_L
the sign of Q_H and Q_L will depend on the device as well

Conservation of energy: $Q = W$ (since no substance is
flowing in or out): net heat in = net work out
net heat out = net work in

First law

For refrigerator or heat pump devices:

$\dot{Q}_H - \dot{Q}_L = \dot{W}_{in}$ First law HP, R

For heat engines $\dot{Q}_H - \dot{Q}_L = \dot{W}$ First law HE

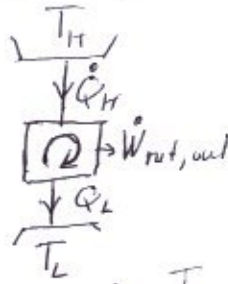
Efficiency

efficiency = $\frac{\text{what you get}}{\text{what you "pay" for it}}$

Heat engine thermal efficiency

$$\eta_{th} = \frac{\dot{W}_{net,out}}{\dot{Q}_H} = \frac{\dot{Q}_H - \dot{Q}_L}{\dot{Q}_H} \quad \text{HE}$$

heat engine



Kelvin-Planck statement of the second law:

\dot{Q}_L must be positive for a heat engine

$$\rightarrow \eta_{th} < 1$$



Refrigerator efficiency

$$\text{COP}_R = \frac{\dot{Q}_L}{\dot{W}_{net,in}} = \frac{\dot{Q}_L}{\dot{Q}_H - \dot{Q}_L}$$



$$\text{COP}_{HP} = \text{COP}_R + 1$$

Heat pump efficiency

$$\text{COP}_{HP} = \frac{\dot{Q}_H}{\dot{W}_{net,in}} = \frac{\dot{Q}_H}{\dot{Q}_H - \dot{Q}_L}$$

Can be greater than 1:
Resistance heater, radiation heater $\text{COP} \approx 1$

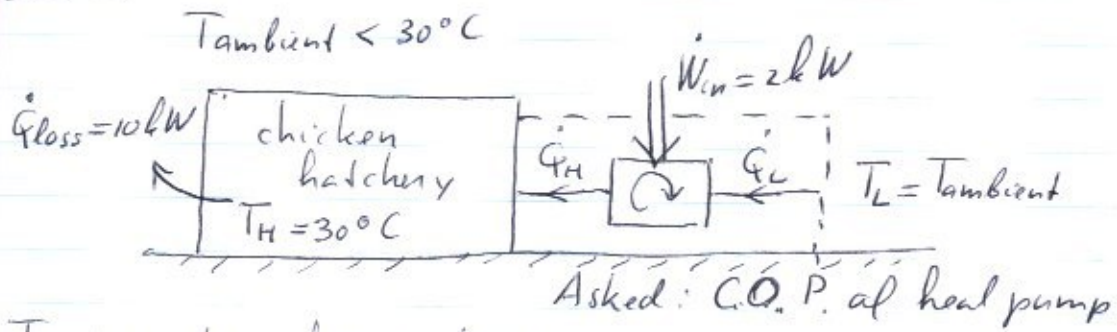
Clausius statement of the second law:

$\dot{W}_{net,in}$ must be positive for a heat pump / refrigerator

Kelvin-Planck & Clausius statements are equivalent:

- 1) If Clausius could be violated you could ~~reverse~~ send \dot{Q}_L from HE back to T_H
- 2) If K.P. could be violated, it acts as a free refrigerator

P 7.22

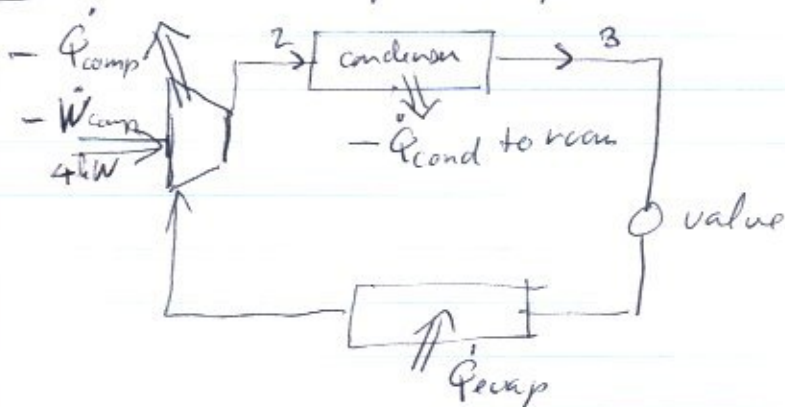


To maintain temperature,

$$\dot{Q}_H = \dot{Q}_{\text{loss}} = 10 \text{ kW}$$

$$\text{Needed C.O.P.} : \beta' = \frac{\dot{Q}_H}{\dot{W}_{\text{in}}} = \frac{10 \text{ kW}}{2 \text{ kW}} = 5$$

7.20 R-12 heat pump p. 6.106



Asked: C.O.P.

Question: What is COP?

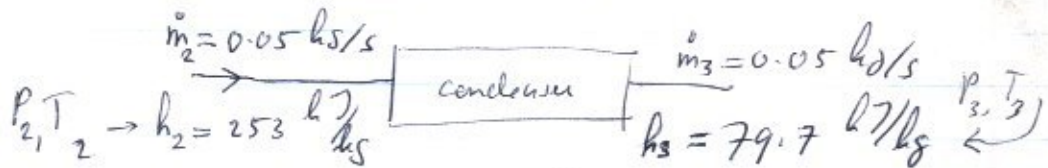
Works to heat room $\rightarrow \beta' = \frac{\dot{Q}_H}{\dot{W}_{net, in}}$

$$\dot{Q}_H \leftarrow -\dot{Q}_{cond}$$

$$\dot{W}_{net, in} \leftarrow -\dot{W}_{comp}$$

$-\dot{W}_{comp}$ is given as 4 kW

To get $-\dot{Q}_{cond}$, use Chapter 6



$$\dot{Q}_{cond} + \dot{m}_2 h_2 = \dot{W} + \dot{m}_3 h_3$$

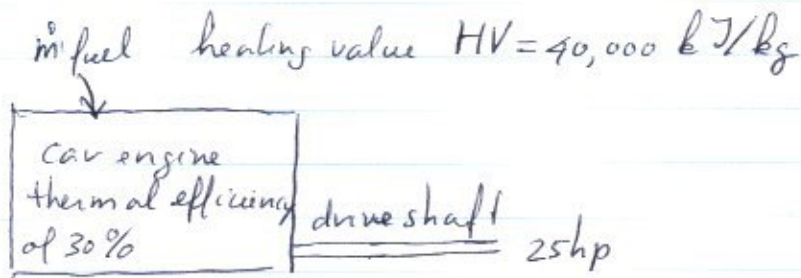
$$\dot{Q}_{cond} = 0.05 \text{ kg/s} (79.7 - 253) \text{ kJ/kg}$$

$$= -8.665 \text{ kW}$$

$$\beta' = \frac{8.665 \text{ kW}}{4 \text{ kW}} = \underline{\underline{2.166}}$$

(This assumes $-\dot{Q}_{comp}$ does not go into the room)

7.24



Asked: rate of fuel consumption (ROFC)
combined power rejected through the radiator and exhaust

Question: What is ROFC?

Question: What is "thermal efficiency of 30%"?

$$\eta_{TH} = \frac{W_{out}}{Q_H} = 0.30$$

Question: What does HV mean?

$$Q_H = \dot{m}_{fuel} HV \Rightarrow 0.30 = \frac{W_{out}}{\dot{m}_{fuel} HV}$$

$1 \text{ hp} = 0.746 \text{ kW}$
 $1 \text{ hp} = 0.7457 \text{ kW}$
 (Table A1: $1 \text{ hp} = 0.7355 \text{ kW}$
 $1 = \frac{0.7355 \text{ kW}}{1 \text{ hp}}$)

Question: What is W_{out} ?

$$W_{out} = 25 \text{ hp} \cdot 0.7355 \frac{\text{kW}}{\text{hp}}$$

Can now compute \dot{m}_{fuel}

$$0.30 = \frac{25 \text{ hp} \cdot 0.7355 \frac{\text{kW}}{\text{hp}}}{\dot{m}_{fuel} 40,000 \frac{\text{kJ}}{\text{kg}}} \Rightarrow \dot{m}_{fuel} = \underline{\underline{0.00153 \text{ kg/s}}}$$

Question: What is CPR TTRAE?

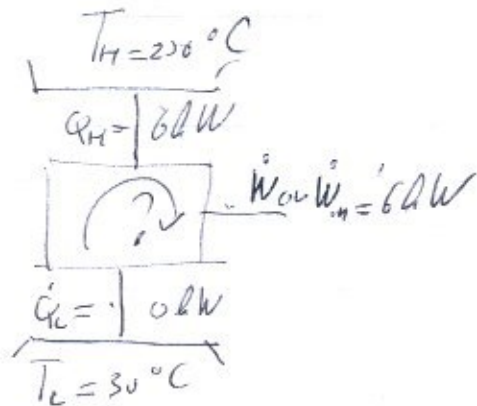
Q_L (very debatable) $\left\{ \begin{array}{l} \rightarrow \text{energy also goes into intake} \\ \rightarrow \text{alternator} \\ \rightarrow \text{power steering} \\ \rightarrow \text{etc.} \end{array} \right.$

Anyway: energy Q_L not converted into drive shaft work:

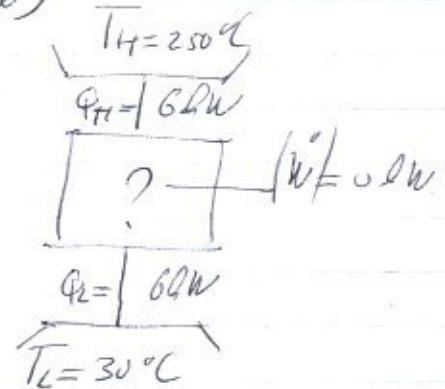
$$Q_L = \dot{m}_{fuel} HV - \dot{W} = 0.00153 \cdot 40,000 \frac{\text{kJ}}{\text{s}} - 25 \text{ hp} \cdot 0.7355 \frac{\text{kW}}{\text{hp}} = 42.8 \text{ kW}$$

7.35]

a)



b)



About a): 1st law is certainly satisfied.
 If it is a heat engine, K.P. is violated: $\dot{Q}_C = 0$
 If it is a heat pump, O.K. $\beta' = 1 = \text{COP}_{HP}$
 A.C. O.K. $\beta = 0 = \text{COP}_R$

About b): 1st law is certainly satisfied.
 If it is a heat ~~engine~~ pump or A.C., violates Clausius
 since work is needed to run it.

Sturab end